

# DENSITY

| Density of Various Solids |                           |
|---------------------------|---------------------------|
| Solid                     | Grams per cm <sup>3</sup> |
| balsa                     | 0.11–0.14                 |
| bone                      | 1.7–2.0                   |
| cement, set               | 2.7–3.0                   |
| cork                      | 0.22–0.26                 |
| diamond                   | 3.01–3.52                 |
| glass, common             | 2.4–2.8                   |
| granite                   | 2.64–2.76                 |
| ice                       | 0.917                     |
| sugar                     | 1.59                      |
| wax, sealing              | 1.8                       |

| Density of Various Liquids |                           |
|----------------------------|---------------------------|
| Liquid                     | Grams per cm <sup>3</sup> |
| acetone (at 20° C)         | 0.792                     |
| ethyl alcohol (at 20° C)   | 0.791                     |
| gasoline                   | 0.66–0.69                 |
| glycerin (at 0° C)         | 1.260                     |
| kerosene                   | 0.82                      |
| mercury                    | 13.6                      |
| olive oil (at 15° C)       | 0.918                     |
| seawater (at 15° C)        | 1.025                     |
| turpentine (spirits)       | 0.87                      |
| water (at 4° C)            | 1.00                      |

Which solid is the most dense?

Which solid is the least dense?

Which liquid is the most dense?

Which liquid is the least dense?

Are all solids more dense than all liquids?

Look at the density of water. Now look at the density of some of the solids. What happens when you put cement in water? What happens when you put cork in water? Why?